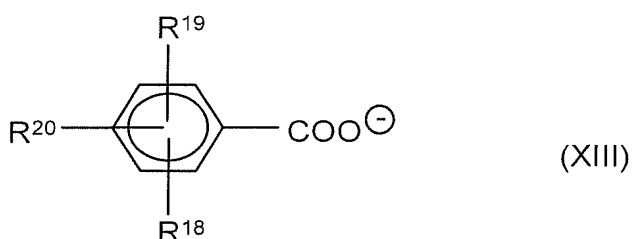
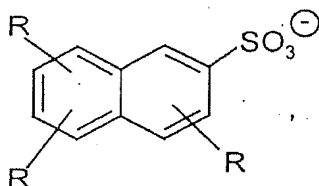
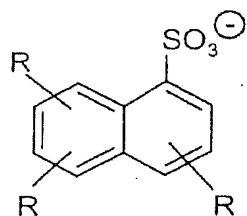
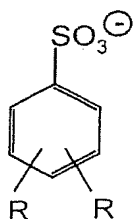
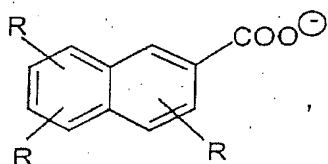
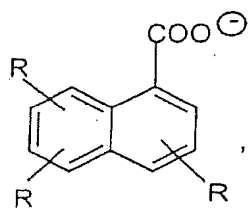


Amendments to the Claims

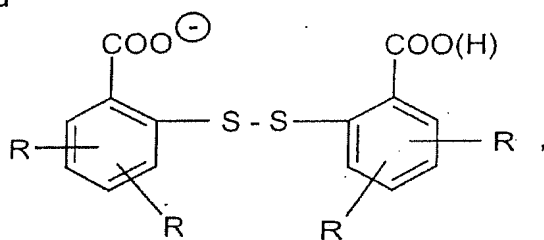
1. (Currently Amended) A process for controlling the charge of an electrophotographic toner, ~~electrophotographic developer, powder, powder coating material, electret material or a chargeable material in an electrostatic separation process~~ comprising the step of adding at least one charge control agent to the electrophotographic toner, ~~electrophotographic developer, powder, powder coating material, electret material or the chargeable material in an electrostatic separation process~~, wherein the at least one charge control agent is a layered double hydroxide salt and wherein the double hydroxide salt is selected from the group consisting of calcined hydrotalcites, uncalcined hydrotalcites, and mixtures thereof having metal cations comprising divalent metal cations and trivalent metal cations, and organic anions A of the formulae (XIII)



wherein R<sup>18</sup>, R<sup>19</sup> and R<sup>20</sup> are identical or different and are hydrogen, C<sub>1</sub>-C<sub>22</sub>-alkyl, C<sub>1</sub>-C<sub>18</sub>-alkenyl, C<sub>1</sub>-C<sub>18</sub>-alkoxy,



and



wherein R is hydrogen,  $\text{C}_1\text{-C}_4\text{-alkyl}$ ,  $\text{C}_1\text{-C}_4\text{-alkoxy}$  or halogen.

2. (Previously Presented) The process as claimed in claim 1, wherein the organic anions are anions from an acid selected from the group consisting of benzoic acid, naphthoic acid, 4-tert-butylbenzoic acid, benzenesulfonic acid, p-toluenesulfonic acid, naphthalenesulfonic acid, and 2,2'-dithiobenzoic acid.
3. (Previously Presented) The process as claimed in claim 1, wherein the number of hydroxyl groups is from about 1.8 to 2.2 times the sum of all the metal cations.
4. (Canceled)
5. (Previously Presented) The process as claimed in claim 1, wherein the at least one double hydroxide salt contains  $\text{Mg}^{2+}$  and  $\text{Al}^{3+}$ .
6. (Previously Presented) The process as claimed in claim 5, wherein the molar ratio  $\text{Mg}^{2+} : \text{Al}^{3+}$  is from 3.1:1 to 1:2.
7. (Previously Presented) The process as claimed in claim 1, wherein the at least one double hydroxide salt is a calcined hydrotalcite.

8. (Previously Presented) The process as claimed in claim 1, wherein the adding step further comprises adding one or more charge control agents selected from the group consisting of triphenylmethanes, ammonium compounds, immonium compounds, iminium compounds, fluorinated ammonium compounds, fluorinated immonium compounds, biscationic acid amides, polymeric ammonium compounds, diallylammonium compounds, aryl sulfide derivatives, phenol derivatives, phosphonium compounds, fluorinated phosphonium compounds, calixarenes, cyclically linked oligosaccharides, derivatives of cyclically linked oligosaccharides, interpolyelectrolyte complexes, polyester salts, metal complex compounds, salicylate-nonmetal complexes, salts of ionic structured silicates, hydroxycarboxylic acid-metal complexes, hydroxycarboxylic acid-nonmetal complexes, benzimidazolones; azines, thiazines and oxazines.

9. (Currently Amended) The process as claimed in claim 1, wherein the at least one charge control agent is present in a concentration of from 0.01% to 50% by weight, based on the total weight of the electrophotographic toner, ~~electrophotographic developer, powder, powder coating material, electret material or chargeable material for electrostatic separation.~~

10. (Currently Amended) An electrophotographic toner, ~~powder or powder coating material,~~ containing from 30% to 99.99% by weight of a binder, from 0.01%

to 50% by weight of at least one layered double hydroxide salt as set forth in claim 1, and, optionally, from 0.001% to 50% by weight of a colorant, based on the total weight of the electrophotographic toner, ~~powder or powder coating material~~.

11. (Previously Presented) The process as claimed in claim 1, wherein the divalent metal cations are selected from the group consisting of  $Mg^{2+}$ ,  $Ca^{2+}$ ,  $Zn^{2+}$ ,  $Co^{2+}$ ,  $Ni^{2+}$ ,  $Fe^{2+}$ ,  $Cu^{2+}$  and  $Mn^{2+}$ .

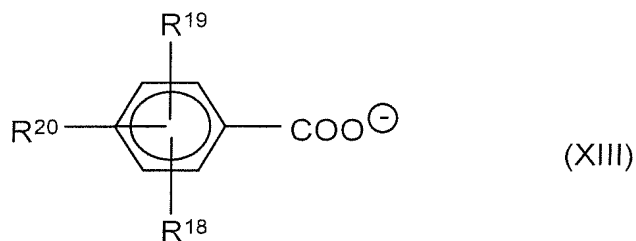
12. (Previously Presented) The process as claimed in claim 1, wherein the trivalent metal cations are selected from the group consisting of  $Al^{3+}$ ,  $Fe^{3+}$ ,  $Co^{3+}$ ,  $Mn^{3+}$ ,  $Ni^{3+}$ ,  $Cr^{3+}$  and  $B^{3+}$ .

13. (Previously Presented) The process as claimed in claim 8, wherein the metal complex compound is a salicylate metal complex.

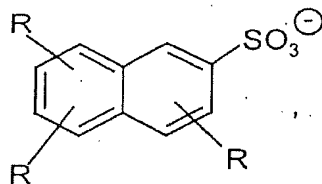
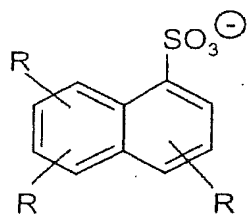
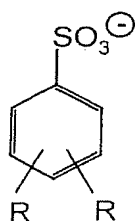
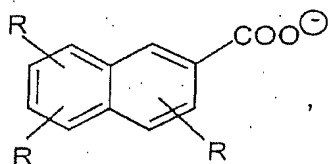
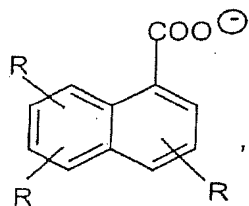
14. (Currently Amended) An electrophotographic toner, ~~electrophotographic developer, powder, powder coating material, electret material or chargeable material for an electrostatic separation process~~ made in accordance with the process of claim 1.

15. (Currently Amended) A process for controlling the charge of an electrophotographic toner, ~~electrophotographic developer, powder, powder coating~~

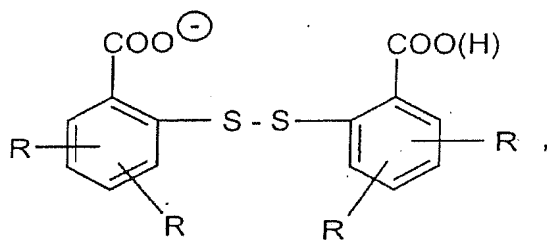
~~material, electret material or a chargeable material in an electrostatic separation~~  
~~process~~ comprising the step of adding at least one charge control agent to the  
electrophotographic toner, ~~electrophotographic developer, powder, powder coating~~  
~~material, electret material or the chargeable material in an electrostatic separation~~  
~~process~~, wherein the at least one charge control agent is a layered double hydroxide  
salt and wherein the double hydroxide salt containing  $\text{Mg}^{2+}$  and  $\text{Al}^{3+}$  wherein the  
molar ratio  $\text{Mg}^{2+} : \text{Al}^{3+}$  is from 3.1:1 to 1:2, and  
organic anions A of the formulae (XIII)



wherein  $\text{R}^{18}$ ,  $\text{R}^{19}$  and  $\text{R}^{20}$  are identical or different and are hydrogen,  $\text{C}_1\text{-C}_{22}$ -alkyl,  
 $\text{C}_1\text{-C}_{18}$ -alkenyl,  $\text{C}_1\text{-C}_{18}$ -alkoxy;



and,



wherein  $\text{R}$  is hydrogen,  $\text{C}_1\text{-C}_4\text{-alkyl}$ ,  $\text{C}_1\text{-C}_4\text{-alkoxy}$  or halogen.

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16 (Canceled)